## Q2. Sources, atoms, and spectra

## A. Light source

In its proper reference frame, a point source emits light in the form of a divergent conical beam, with the angular width of $90^{\circ}$ (from $-45^{\circ}$ to $+45^{\circ}$ with respect to the cone axis). In a reference frame which moves towards the source with an unknown speed $v$, the angular width of the beam is of only $60^{\circ}$ (from $-30^{\circ}$ to $+30^{\circ}$ with respect to the same cone axis). The light speed in vacuum is $\mathrm{c}=2.998 \cdot 10^{8} \frac{\mathrm{~m}}{\mathrm{~s}}$.

| A | Determine the speed $v$ of the source. | 2.50 p. |
| :--- | :--- | :--- |

## B. Balmer emission spectrum

The spectral resolving power of a spectrometer is $R=5 \cdot 10^{5}$. The spectrometer is used to observe the Balmer series in the emission spectrum of the hydrogen atom (the visible domain).

Note: The possible mechanisms of broadening of the spectral lines (Lorentzian, Gaussian, etc.) will not be considered.

| B.1 | Express the mathematical definition of the spectral resolving power of the <br> instrument. | 0.25 p. |
| :--- | :--- | :--- |

B. 2 Determine the highest value for the principal quantum number $n$ of the energy level for which the spectral line emitted by an atom for the transition to the level $n^{\prime}=2$ can still be distinctly resolved by the instrument, with 2.25 p . respect with its neighbours.

## C. Absorption spectra

The energy levels of an atom are given by $E_{n}=-\frac{A}{n^{2}}$, where $n$ is an integer and $A$ is a positive constant. Among the adjacent spectral lines which, at room temperature, the atom can absorb, two have the wavelengths 97.5 nm and 102.8 nm , respectively. The elementary electric charge is $e=1.602$. $10^{-19} \mathrm{C}$, the speed of light in vacuum is $\mathrm{c}=2.998 \cdot 10^{8} \frac{\mathrm{~m}}{\mathrm{~s}}$, and Planck's constant is $h=6.626$. $10^{-34} \mathrm{~J} \cdot \mathrm{~s}$.

| C. 1 | Find the values of the quantum numbers n of the energy levels implied in <br> the transitions. | 3.00 p. |
| :--- | :--- | :--- |

$\mid$ C. 2 Determine the value of the constant A in joule and in electron-volt. $\quad 1.50 \mathrm{p} .7$.

© prof. Florea ULIU, PhD, University of Craiova

## Q3. Sources, atoms, and spectra

## Answer sheet

| $\mathbf{A}$ | Final expression for the speed of the <br> source | Numerical value for the speed of <br> the source |  |
| :--- | :--- | :--- | :--- |
|  | $v=$ | $v=$ | 2.50 p. |


| B.1 |  |  |
| :--- | :--- | :--- |
|  | $R=$ | 0.25 p. |


| B. 2 |  |  |
| :--- | :--- | :--- |
|  | $n=$ | 2.25 p. |


| C. 1 |  |  |
| :--- | :--- | :--- |
|  | $n=$ | 3.00 p. |


| C.2 | in joule | $A=$ | 1.00 p. |
| :--- | :--- | :--- | :--- |
|  | in electron-volt | $A=$ | 0.50 p |


| C. 3 |  | 0.50 p. |
| :--- | :--- | :--- |

[^0]
[^0]:    Theoretical Problem $\mathcal{N}$ o. 3

